



Vocabulary

Define and know the meaning of each of the following terms.

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|------------------------|--------------------------------|-----------------------|-------------------------|---------------------------------|
| 1. aqueous | 7. decomposition reaction | 12. exothermic | 18. percent composition | 23. single replacement reaction |
| 2. balancing equations | 8. double replacement reaction | 13. formula mass | 19. precipitate | 24. stoichiometry |
| 3. catalyst | 9. empirical formula | 14. gram formula mass | 20. products | 25. synthesis reaction |
| 4. chemical equation | 10. endothermic | 15. mole | 21. reactants | |
| 5. chemical formula | 11. end reaction | 16. mole ratio | 22. reversible reaction | |
| 6. coefficient | | 17. molecular formula | | |



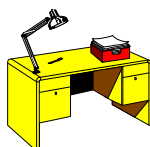
Reading and Study

ASSIGNMENT

Chapter # 2
Chapter # 3
Chapter # 9

SECTIONS

2.1
3.2
9.1-9.2



Homework Reading and Questions

READING		REVIEW QUESTIONS/PAGE
HW#29	Read sec 9.2 p. 276	Q9-11 p. 301
HW#30	Read sec 2.1 p. 43-45	Q3-5 p. 64
HW#31	Read sec 3.2 p. 71-73	-
HW#32	Read sec 9.1 p. 271-274	Q1-6 p. 301
HW#33	Read sec 9.2 p. 277-279	Q12-15 p. 301
HW#34	Read sec 9.2 p. 274-275	-