



Vocabulary

Define and know the meaning of each of the following terms.

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|------------------------|----------------------------------|---------------------------|---------------------|--------------------------|
| 1. anion | 10. conductivity | 18. flexible | 27. network solid | 36. symmetrical |
| 2. asymmetrical | 11. covalent bond | 19. hydrogen bond | 28. oxidation state | 37. univalent |
| 3. binary covalent | 12. cross over rule | 20. intermolecular forces | 29. polar molecule | 38. valence electrons |
| 4. bond | 13. diatomic molecule | 21. ion | 30. polar bond | 39. valence shell |
| 5. bond angle | 14. dipole | 22. ionic bonding | 31. polyatomic ion | 40. Van der Waals Forces |
| 6. cation | 15. dipole-dipole attraction | 23. malleable | 32. polyvalent | |
| 7. chemical bond | 16. ductile | 24. metallic bond | 33. stable octet | |
| 8. charge distribution | 17. electronegativity difference | 25. molecule | 34. stock system | |
| 9. coefficient | | 26. monatomic | 35. subscript | |



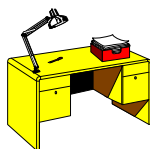
Reading and Study

ASSIGNMENT

Chapter # 2
Chapter # 6
Chapter # 7
Chapter #18

SECTIONS

2.3
6.1-6.4, 6.6-6.7
7.1
18.4



Homework Reading and Questions

READING	REVIEW QUESTIONS / PAGE
HW#19 Read sec 6.1-6.3 p. 173-182	Q7-11 p. 200
HW#20 -	Q15-17 p. 201
HW#21 Read sec 6.4 p. 182-186;	Q20-24 p. 201
HW#22 Read sec 6.6 -6.7 p. 191-198	Q31-33 p. 201
HW#23 -	Q30 & 34 p. 201
HW#24 -	Q10 p. 64
HW#25 Read sec 2.3 p. 49 -50	Q12 p. 64
HW#26 Read p. 49-50	Q13 p. 64
HW#27 Read sec 18.4 p. 599-603	-
HW#28 Read sec 7.1 p. 206-212	Q 1-6 p. 231