



Vocabulary

Define and know the meaning of each of the following terms.

- | | | | | |
|---------------------|---------------------------|------------------------------|----------------------------|--------------------|
| 1. boiling point | 10. freezing | 19. Kelvin scale | 27. phase | 36. sublimation |
| 2. Boyle's Law | 11. freezing point | 20. kilopascal | 28. phase change | 37. temperature |
| 3. Celsius scale | 12. gas | 21. liquid | 29. phase diagram | 38. torr |
| 4. Charles' Law | 13. heat | 22. kinetic molecular theory | 30. pressure | 39. vapor |
| 5. combined gas law | 14. heat of fusion | 23. melting point | 31. solid | 40. vapor pressure |
| 6. condensation | 15. heat of vaporization | 24. normal boiling point | 32. specific heat capacity | 41. water vapor |
| 7. deposition | 16. heating/cooling curve | 25. normal melting point | 33. standard pressure | |
| 8. energy | 17. ideal gas | 26. one atmosphere | 34. standard temperature | |
| 9. evaporation | 18. joule | | 35. STP | |



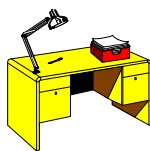
Reading and Study

ASSIGNMENT

Chapter # 2
Chapter #14

SECTIONS

2.2
14.2-14.4



Homework Questions

READING	QUESTIONS / PAGE
HW#44 Read p. 61- 64	Q1-3 p. 64
HW#45 Read Sec 14.2-14.3; p. 467-478	Q1 p. 470; Q2 p.478
HW#46 Read Sec 14.4 p. 479-488	Q1-2, 4-5 p. 489