



Vocabulary

Define and know the meaning of each of the following terms.

- | | | | | |
|------------------------|--------------------------------|-----------------------|-------------------------|---------------------------------|
| 1. aqueous | 7. decomposition reaction | 12. exothermic | 18. percent composition | 23. single replacement reaction |
| 2. balancing equations | 8. double replacement reaction | 13. formula mass | 19. precipitate | 24. stoichiometry |
| 3. catalyst | 9. empirical formula | 14. gram formula mass | 20. products | 25. synthesis reaction |
| 4. chemical equation | 10. endothermic | 15. mole | 21. reactants | |
| 5. chemical formula | 11. end reaction | 16. mole ratio | 22. reversible reaction | |
| 6. coefficient | | 17. molecular formula | | |



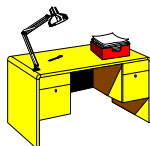
Reading and Study

ASSIGNMENT

Chapter # 2
Chapter # 9
Chapter #10
Chapter #11

SECTIONS

2.3
9.1-9.3
10.1-10.3
11.1-11.2



Homework Reading and Questions

READING	QUESTIONS/PAGE
HW#34 Read p. 312-313	-
HW#35 Read p. 332-334	-
HW#36 Read p. 68-71	-
HW#37 Read p. 279-284	-
HW#38 Read p. 291-297	Q1-4 p. 297
HW#39 Read p. 284-290	Q3 p. 290
HW#40 Read p. 313-322 and p. 323-331	Q1-3 p. 322 ; Q1-2 p. 331
HW#41 Read p. 347-352	Q1-3 p. 353
HW#42 Read p.335-339	Q3 p. 339
HW#43 Read p. 354-357	Q3 p. 364